

Low-Energy Solvents for CO₂ Capture Enabled by a Combination of Enzymes and Vacuum Regeneration

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AICHE 2014 meeting
Atlanta, GA
November 17, 2014

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Agenda



- Project Overview
 - Carbonic anhydrase for CO₂ capture
 - Partners, budget & objective
- Technology Background
 - Process concept
 - Fundamental mechanism
- Progress and Status
 - Bench-scale system description
 - Parametric test results
 - 500-h run initial test results
- Conclusions & Next Steps

Carbonic anhydrase for CO₂ capture: a select survey of other public projects

■ Codexis

- DE-AR0000071
- PE dissolved CA for 4.2 M MDEA and >85°C
- Stable performance, 60 h test at NCCC

■ Akermin

- DE-FE0004228
- Immobilize CA in absorber
- Stable performance, 116 days at NCCC

■ Illinois State Geological Survey, Prairie Research Institute, Yongqi Lu Laboratory

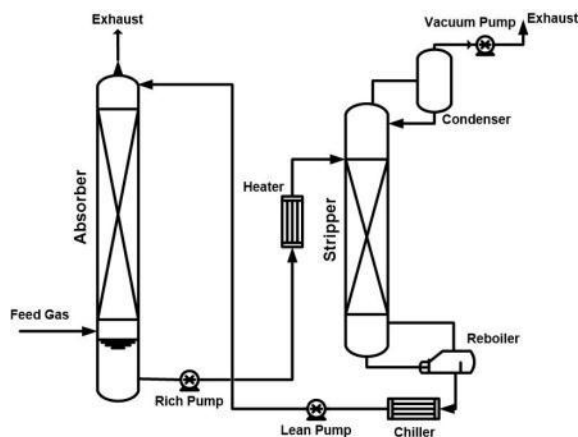
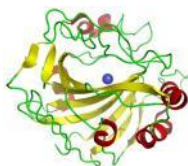
- DE-FC26-08NT0005498 and others
- Evaluate process concepts, immobilization options, temp. stability, kinetics, etc.

About the NCCC:

- The National Carbon Capture Center provides a slipstream from a 880 MW coal unit for testing post-combustion CO₂ capture technologies.

Project Objective

Complete a *bench-scale study* and corresponding full technology assessment to validate the potential in meeting the DOE Program Objectives of a *solvent-based post-combustion carbon dioxide capture* system that integrates



- a **low-enthalpy**, aqueous potassium carbonate-based solvent
- with an **absorption**-enhancing (*dissolved*) carbonic anhydrase enzyme catalyst
- and a low temperature vacuum **regenerator**
- in a **re-circulating** absorption-desorption process configuration

Novozymes in Brief – World Leader in Bioinnovation

Producing large volume enzymes for industrial applications

1. Improving the production host

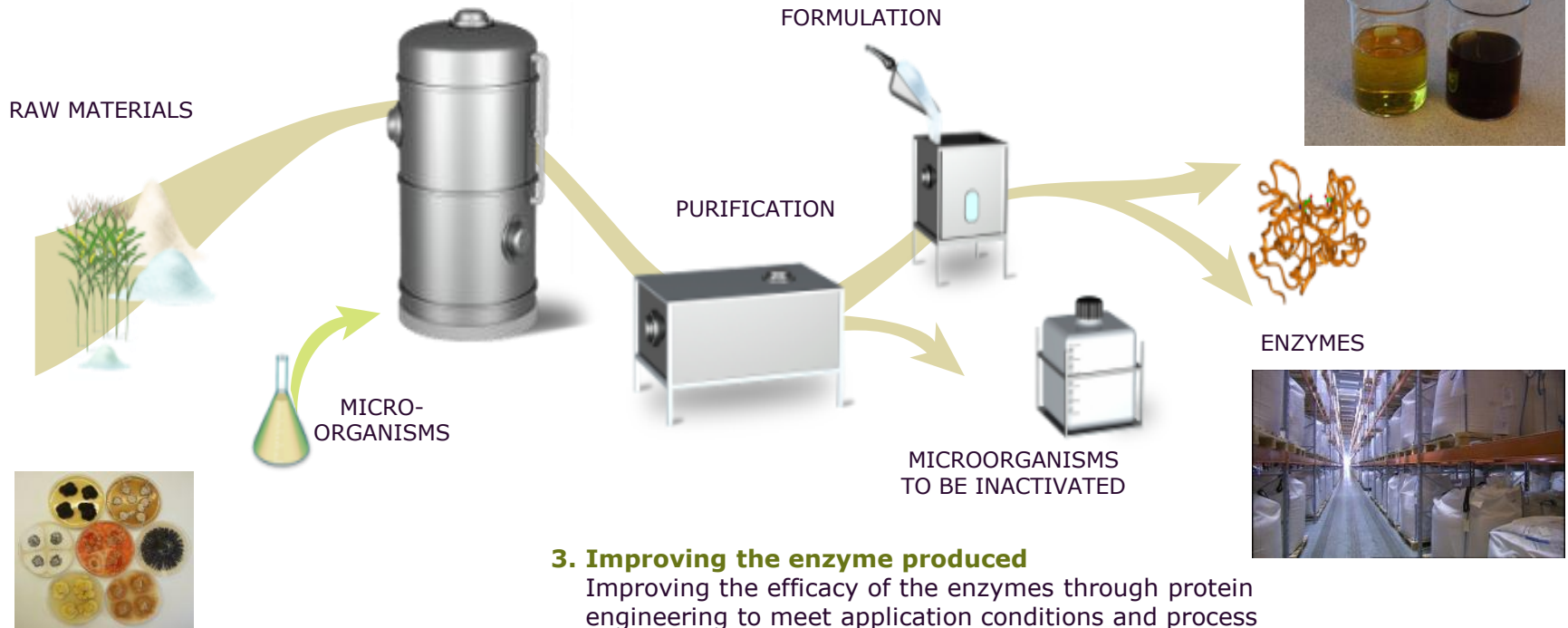
Improving the microorganisms' ability to produce more enzymes per m³ fermentation tank through genetic engineering



FERMENTATION

2. Optimizing industrial production

- Process optimization
- Equipment optimization
- Input optimization



3. Improving the enzyme produced

Improving the efficacy of the enzymes through protein engineering to meet application conditions and process economy requirements

Project Overview

■ Project Participants



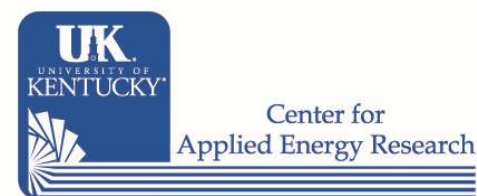
Ultrasonics & Aspen®



Full Process Analysis



Enzymes & Solvents



Kinetics & Bench-scale Tests

- DOE Project Manager: Andrew Jones
- Project Number: DE-FE0007741
- Total Project Budget: \$2,088,644
 - DOE: \$1,658,620
 - Cost Share: \$430,024
- Project Duration: Oct. 1, 2011 – March 31, 2015

DOE Program Objectives

Develop solvent-based, post-combustion technology that

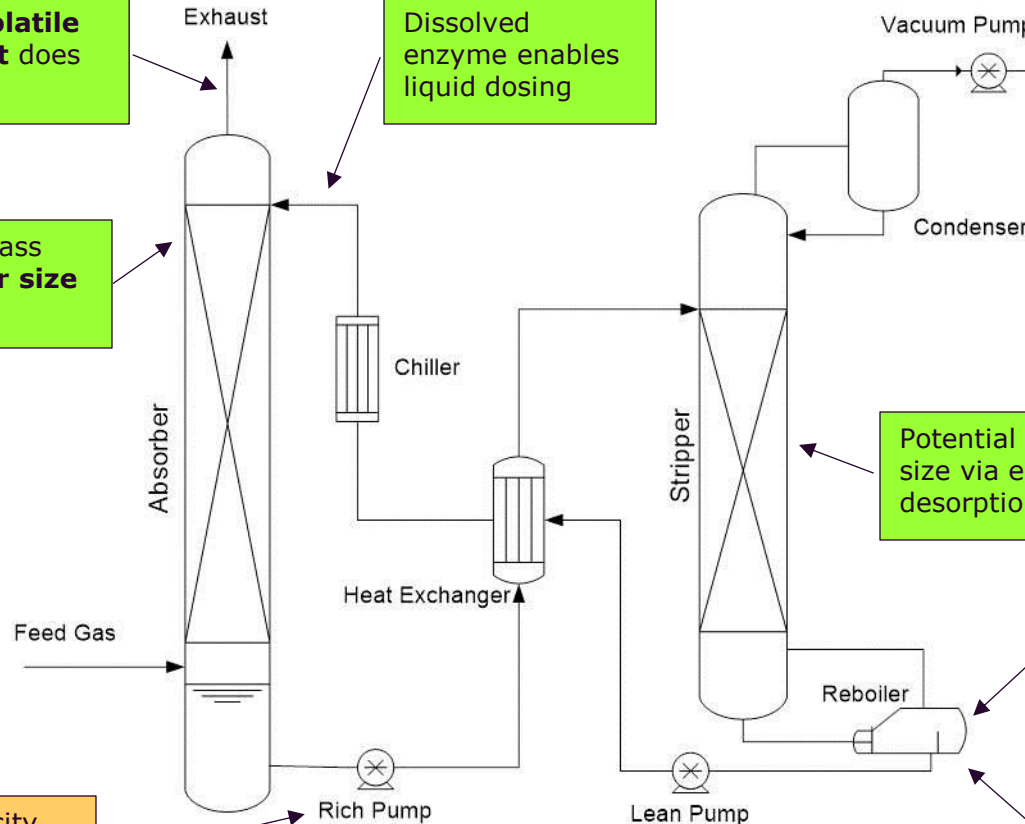
- Can achieve $\geq 90\%$ CO₂ removal from coal-fired power plants
- Demonstrates progress toward the DOE target of <35% increase in LCOE.

Process Concept, Advantages & Challenges

Stable, benign, non-volatile
aq. K_2CO_3 -based **solvent** does
not require water wash

Enzyme-enhanced CO_2 mass
transfer **reduces absorber size**
to feasible height

Dissolved
enzyme enables
liquid dosing



Increased compression
energy to account for
vacuum regen condition

Potential to minimize stripper
size via enzyme-enhanced CO_2
desorption (simulation)

Potential to **use low pressure**
steam in combination with
vacuum for low enthalpy K_2CO_3
regeneration

K_2CO_3 loading capacity
limit may increase
solvent circulation rate

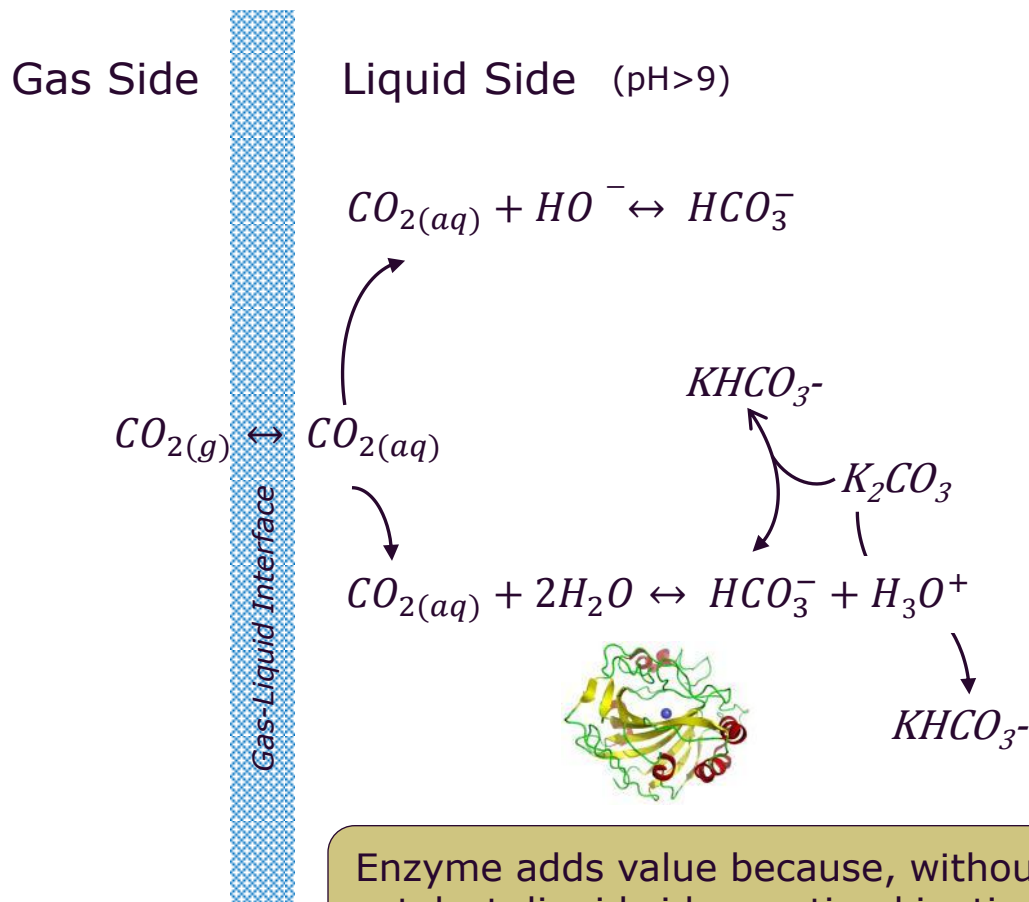
Enzyme temperature limits
may result in **high enzyme**
replenishment requirement

Absorption
1 atm/30-40°C

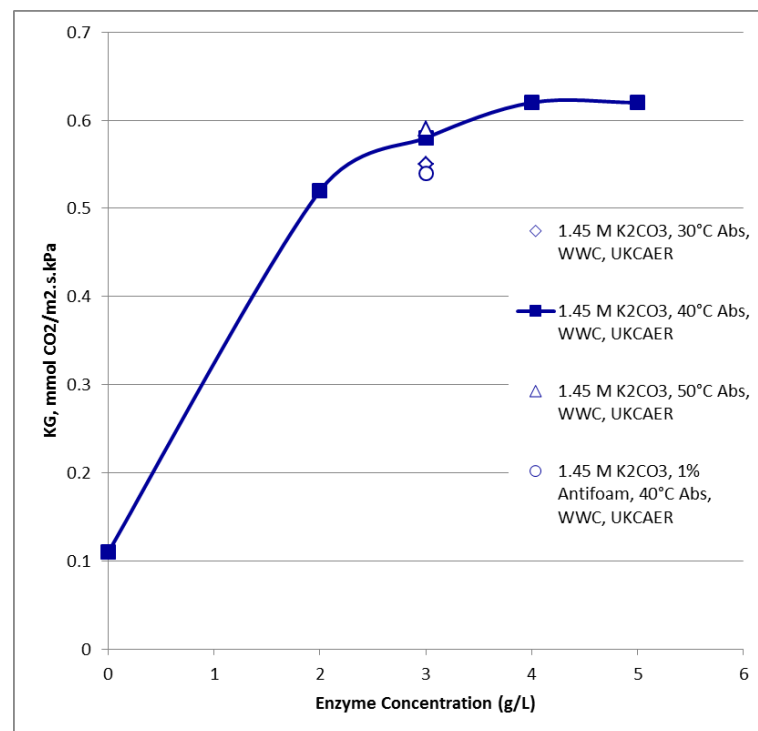
Regeneration
~0.35 atm/76°C

Generating Bench-scale Test Data

Enzyme Enhanced CO₂ Absorption Mechanism



Enzyme adds value because, without catalyst, liquid side reaction kinetics are overall mass transfer rate limiting



Overall Mass Transfer Coefficient (K_g) Enhanced by Enzyme in WWC

Bench-scale Unit Description

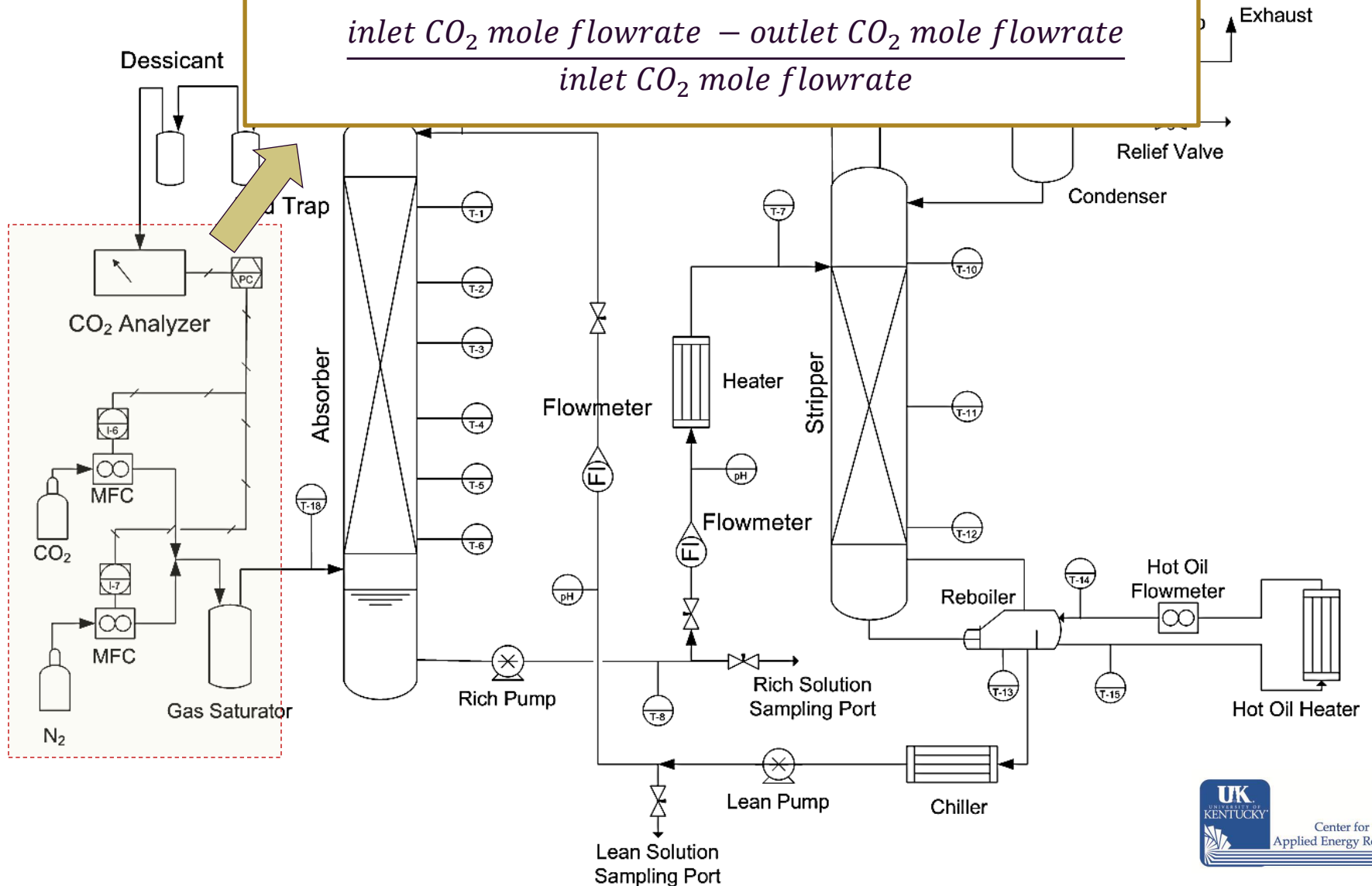


- Flow Rates
 - Gas: 30 SLPM (15% CO₂)
 - Liquid : 300-600 ml/min
- Liquid Temperature
 - Absorber Inlet: 30-40°C
 - Stripper Inlet: ~65°C
 - Reboiler Oil Inlet: 90-95°C
- Stripper Pressure: 0.35 atm absolute
- K₂CO₃ Concentration: 23 wt%
- Enzyme Concentration: 0 – 4 g/L

PFD of Integrated Bench-scale System

Capture Efficiency =

$$\frac{\text{inlet } CO_2 \text{ mole flowrate} - \text{outlet } CO_2 \text{ mole flowrate}}{\text{inlet } CO_2 \text{ mole flowrate}}$$



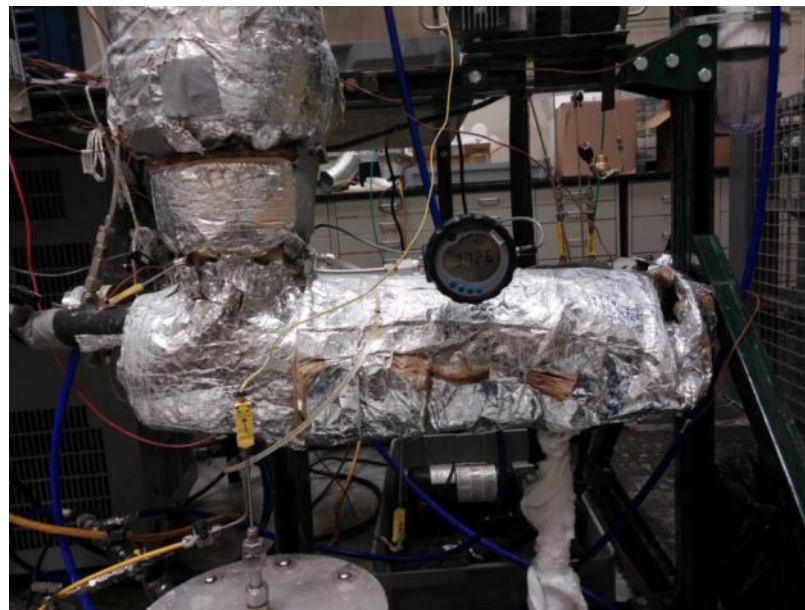
Bench-scale Operational Observations



Absorber bottom



Stripper top



Stripper bottom

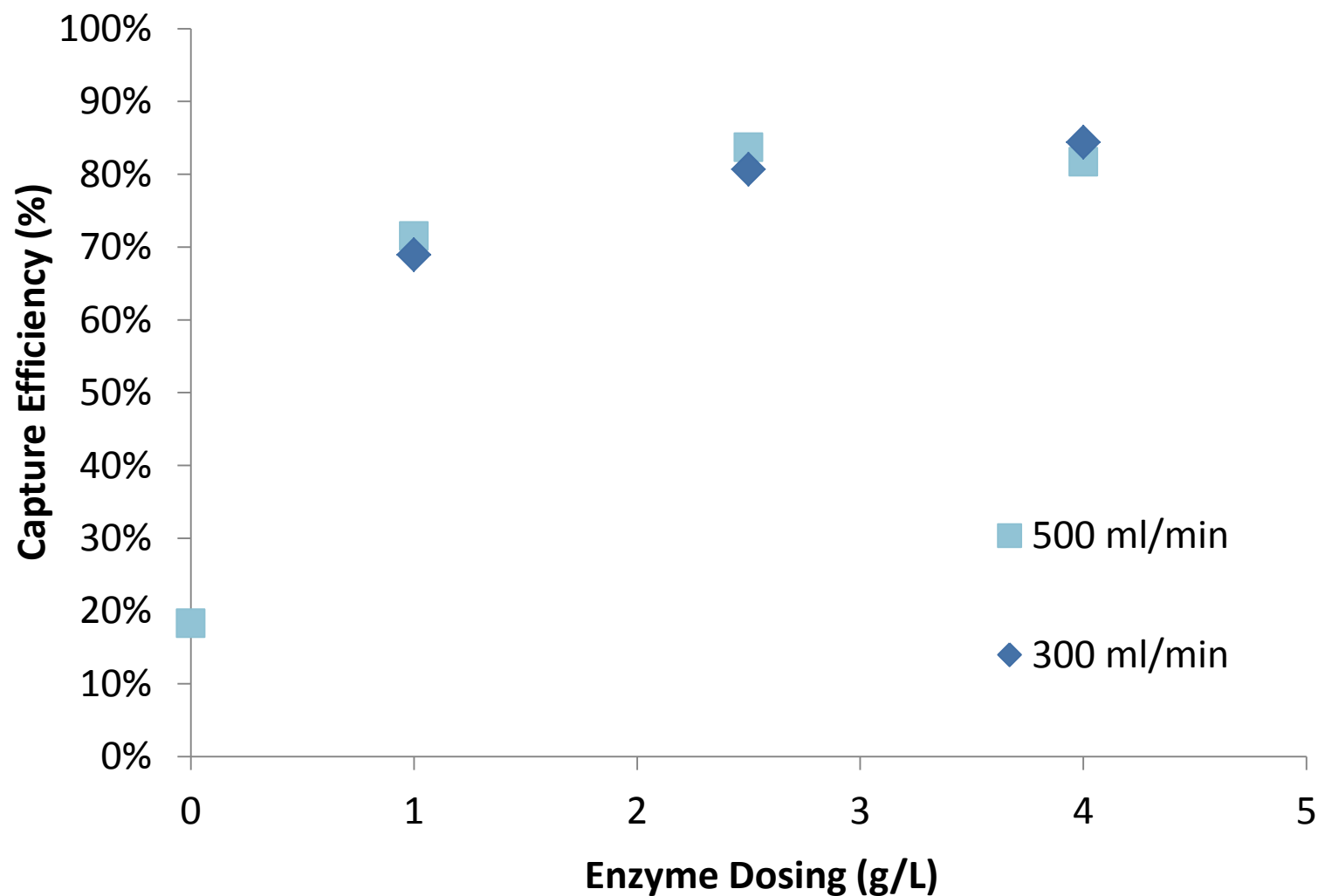
Absorber

- Low temp difference along absorber length due to low enthalpy solvent
- No visual change in packing
- Rich solvent filter removes solids

Stripper

- Water cooled condenser at top
- Tube and shell reboiler
- Antifoam dosing effectively mitigates foaming
- Bulk temp. ranges from 65°C (rich solvent inlet to stripper top) to 76°C (reboiler)
- Reboiler heating source temp. 90-95°C
- Lean solvent filter removes solids

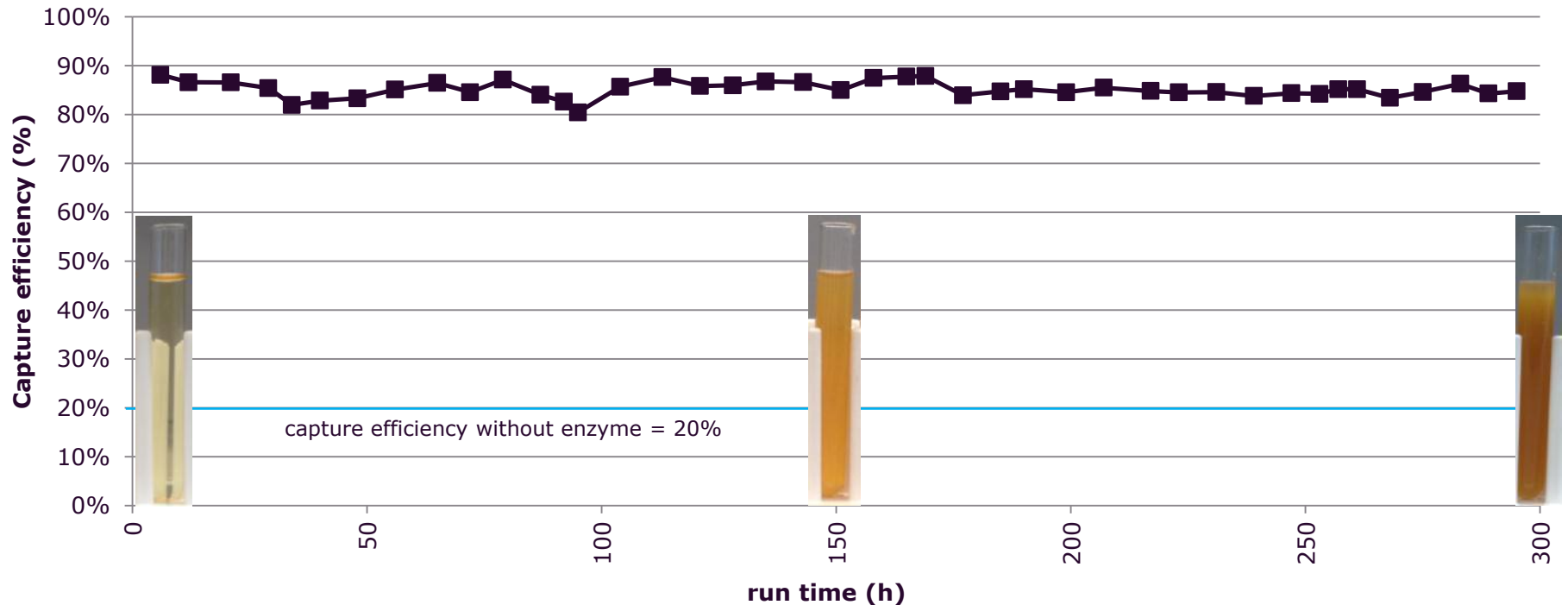
Impact of Enzyme Conc. and Liquid Flow Rate



500 Hour Long Term Test

- Baseline conditions
 - 40°C absorber
 - 95°C reboiler heating source temperature
 - 0.35 atm absolute stripper top pressure
 - 500 ml/min liquid flow rate
 - 30 SLPM gas flow rate; 15% CO₂
 - 2.5 g/L active enzyme dose
 - 23 wt% K₂CO₃
- Solvent additions per ~7 h run day
 - Antifoam addition: ~0.04%
 - Active enzyme addition: ~20% of active enzyme dose
 - Defining enzyme replenishment is part of the program
 - Active enzyme replenishment is at least 7% of total enzyme
 - Solvent volume and alkalinity are maintained

500 Hour Long Term Test Results

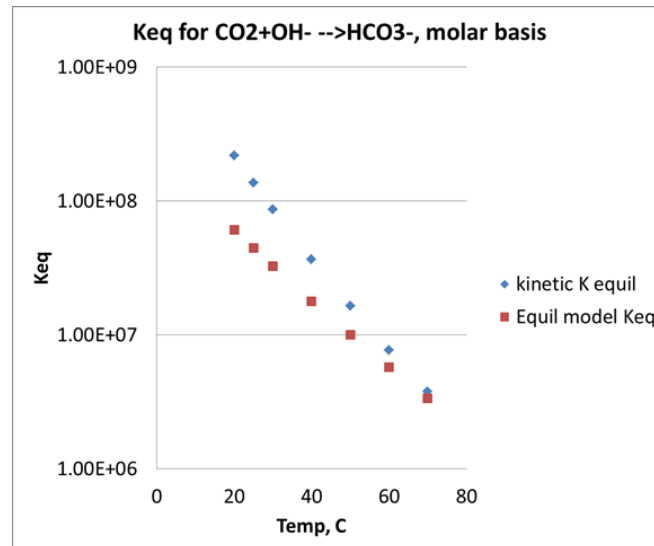


- >80% CO₂ Capture efficiency maintained
- Solvent darkens and becomes turbid
- Principle of using dissolved enzyme replenishment to achieve stable operation demonstrated

Approach to Kinetic Model

- Improve existing ASPEN kinetic model for $\text{CO}_2 + \text{OH}^- \rightarrow \text{HCO}_3^-$
 - Include data representing a wider temperature range than prior model
 - Include the effects of ionic strength on rate
 - Correct existing reverse kinetics to provide agreement with equilibrium model predictions at temperatures $< 70^\circ\text{C}$.
- Include a parallel rate expression for $\text{CO}_2 + 2\text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{HCO}_3^-$
 - Model enzyme effect by accelerating this reaction, not hydroxide reaction

Comparison of equilibrium constants predicted by equilibrium model and pre-correction kinetic model.



Conclusions and Next Steps

▪ Conclusions

- 30 SLPM benchscale unit is operational and providing unique test data for low P/low T stripping with enzyme-enhanced K_2CO_3 -based solvent
- Parametric testing resulted in selection of 500 hour test conditions operating at >80% capture
- Current enzyme longevity is significantly diminished by travel through stripper, but can be mitigated for test purposes by replenishment program

▪ Next Steps

- Complete 500 hour testing
- Prepare full TEA, with updated kinetics-based process simulation and ASPEN models
- Complete EH&S assessment

▪ Potential Future Developments

- Improve enzyme thermal stability
 - Immobilization or chemical modification to create physical barrier to unfolding
 - ID alternate enzyme candidates and/or protein engineering to improve T stability
 - Explore alternate process configurations to reduce enzyme loss
- Evaluate options for increasing liquid loading capacity

Acknowledgements

DOE-NETL

Andrew Jones

Pacific Northwest National Laboratory

Charles Freeman (PM/TL), Mark Bearden
Greg Whyatt

UK-Center for Applied Energy Research

Kunlei Liu (PM), Kun Liu (TL), Guojie Qi
Reynolds Frimpong

Doosan Power Systems

David Fitzgerald (PM), Jonathan Slater (TL)
Rafael Vidal, Tania Russell

Novozymes

Sonja Salmon (PI/PM), Alan House (TL)
Erin Yarborough

